

LETTER TO THE EDITOR

APEX 1 mm line survey of the Orion Bar

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ABSTRACT

Context. Unbiased molecular line surveys are a powerful tool for analyzing the physical and chemical parameters of astronomical objects and are the only means for obtaining a complete view of the molecular inventory for a given source. The present work stands for the first such investigation of a photon-dominated region.

Aims. The first results of an ongoing millimeter-wave survey obtained towards the Orion Bar are reported.

Methods. The APEX telescope in combination with the APEX-2A facility receiver was employed in this investigation.

Results. We derived the physical parameters of the gas through LVG analyses of the methanol and formaldehyde data. Information on the sulfur and deuterium chemistry of photon-dominated regions is obtained from detections of several sulfur-bearing molecules and DCN.

Key words. ISM: individual objects: Orion Bar – ISM: abundances – ISM: molecules

1. Introduction

Photon-dominated regions (PDRs) are commonly found interstellar environments whose global properties are determined by intense far-ultraviolet (FUV) radiation emerging from nearby young OB stars (e.g. Hollenbach & Tielens 1997). Owing to its proximity and nearly edge-on orientation, the Orion Bar PDR has received particular attention as a template for studies of the spatial stratification of various atomic and molecular species from the highly penetrated surface layers deep into the parental molecular cloud. Targeted (sub)millimeter-wave investigations indicate that, besides a considerable number of ubiquitous astronomical molecules such as CO, CS, HCN, CH₃OH, and H₂CO, other species can be found that are suggestive of a unique PDR chemistry (Hogerheijde et al. 1995; Jansen et al. 1995; Fuente et al. 2003). One such example is the molecular ion CO⁺ (e.g. Störzer et al. 1995), and another one appears to be CF⁺, which so far has only been detected toward the Orion Bar (Neufeld et al. 2006, this volume). PDRs are also thought to show enhanced abundances of molecules carrying refractory elements due to grain breakup caused by the FUV radiation field (Schilke et al. 2001). Additionally, observations of the Horsehead nebula suggest that at least part of the molecular carbon chain budget in PDRs may be produced from UV-destruction of PAHs and carbonaceous grains (Pety et al. 2005).

In view of this, PDRs are attractive targets for unbiased molecular line studies, which also help to derive the global picture of the physics and chemistry associated with them. Here, we present the initial results from an unbiased molecular line survey obtained between 279 and 308 GHz with the APEX telescope (Güsten et al. 2006, this volume).

2. Observation

Observations were conducted with the APEX 12-m telescope¹ towards the Orion Bar at the position $\alpha_{2000} = 5^{\text{h}}35^{\text{m}}25.3^{\text{s}}$, $\delta_{2000} = -5^{\circ}24'34.0''$, corresponding to the “Orion Bar (HCN)” position of Schilke et al. (2001), the most massive clump seen in H¹³CN (Lis & Schilke 2003, hereafter LS03). Data were taken in 2005, between October and December in the position-switching mode, with the reference position at (600'', 0''). The APEX-2A receiver (Risacher et al. 2006, this volume) was used in combination with the Fast Fourier Transform Spectrometer (Klein et al. 2006, this volume) providing a bandwidth of 1 GHz and a resolution of 0.12 km s⁻¹. The frequency range between 279 and 307.7 GHz was completely covered; additional data were taken in selected frequency ranges from 318.5 GHz to 361.5 GHz. Step-widths of approximately 500 MHz were used to cover each frequency setup twice to facilitate the sideband assignment. The antenna temperature was converted to the main-beam brightness temperature by using forward and beam efficiencies of 0.97 and 0.74, respectively. The noise level is of the order of 0.06 K–0.1 K, and the system temperature around 160 K. The calibration was performed by using the APECS software (Muders et al. 2006, this volume). The pointing was checked on Mars and was found to be accurate within a few arc-seconds. We estimate a calibration uncertainty of ~30%.

¹ This publication is based on data acquired with the Atacama Pathfinder Experiment (APEX). APEX is a collaboration between the Max-Planck-Institut für Radioastronomie, the European Southern Observatory, and the Onsala Space Observatory.

3. Analysis of the data

The line identification was based on the Cologne Database for Molecular Spectroscopy² (Müller et al. 2001) and on the JPL³ (Pickett et al. 1998) line catalog and performed with the XCLASS software. We identified 16 different species and a number of their isotopologs; no strong unidentified lines were found in the range surveyed. In the following paragraphs, we focus on the excitation conditions of methanol (CH₃OH) and formaldehyde (H₂CO), on the one hand, and on the analysis of S-bearing molecules, on the other. Methanol and formaldehyde are present in our dataset with a large number of transitions (Table 1), and their collisional rates are available (Green 1991; Pottage et al. 2002). Thus, a more rigorous multi-line analysis is possible for these two species, which delivers temperature and density estimates. As for the S-bearing molecules in our dataset, some of them, namely CS, H₂S, SO, SO⁺, and SO₂, have already been studied towards this source (Hogerheijde et al. 1995; Jansen et al. 1995; Fuente et al. 2003), whereas HCS⁺, H₂CS and NS are being analyzed here for the first time in this environment. A comprehensive analysis will be presented once the line survey has been completed.

3.1. CH₃OH and H₂CO

The excitation analysis of CH₃OH and H₂CO was performed by using the method described by Leurini et al. (2004), recently modified to treat H₂CO as well. The procedure is based on the simultaneous fit of all the lines with a synthetic spectrum computed in the LVG approximation with the cosmic background as the only external radiation field. If lines are optically thick, the model fits source size, temperature, H₂ density, and CH₃OH/H₂CO column density; for optically thin lines, source size and column density cannot be determined independently, and beam-averaged column densities are provided.

To improve the signal-to-noise ratio in the weaker transitions, we smoothed the data to a resolution of 0.5 km s⁻¹. At this resolution, each line has a single-peak Gaussian profile ($v_{\text{lsr}} = 10.0 \pm 0.2$ km s⁻¹, $\Delta v = 1.7 \pm 0.3$ km s⁻¹), while, at the original resolution of 0.12 km s⁻¹, two peaks were detected in the 5_{1,5} → 4_{1,4} H₂CO and in the 6₋₁ → 5₋₁ CH₃OH-*E* transitions. Non-Gaussian profiles were found in other transitions as well. By fitting the 6₋₁ → 5₋₁ CH₃OH-*E* line with the GAUSS method of CLASS, we found two velocity components at $v_{\text{lsr}} = 10.89 \pm 0.06$ and $v_{\text{lsr}} = 9.82 \pm 0.05$ km s⁻¹, with $\Delta v = 0.62 \pm 0.13$ and $\Delta v = 0.63 \pm 0.06$ km s⁻¹. These values resemble what LS03 found for the H¹³CN clumps 1, 5 and 8, which are all within our beam. For the sake of simplicity, one velocity component was used both for the CH₃OH and the H₂CO analysis, as derived from the smoothed CH₃OH data.

High density ($n \sim 7 \times 10^6$ cm⁻³) and a moderately high temperature ($T \sim 70$ K) are needed to reproduce the CH₃OH spectrum. The temperature is constrained to this value by the non-detection of high-excitation ($E_{\text{low}} \geq 80$ K) lines (Fig. 1). The high density value we derived suggests that CH₃OH comes from small clumps. Therefore, guided by LS03, we used a source size of 5'' and derived $N(\text{CH}_3\text{OH-}E) = 7 \times 10^{14}$ cm⁻² and $N(\text{CH}_3\text{OH-A}) = 8 \times 10^{14}$ cm⁻². Our estimate of n in the clumps agrees with previous studies (e.g. Young Owl et al. 2000; LS03); T_{kin} is also similar to previous results for the clump gas (see discussion in LS03) and to theoretical models (Gorti & Hollenbach 2002). The

Table 1. Line parameters of the observed transitions.

Transition	Frequency [MHz]	E_{low} [K]	$\int T_{\text{MB}} dv$ [K km s ⁻¹]
H ₂ CO 4 _{1,4} → 3 _{1,3}	281526.929	32.06	10.7 ± 0.5
H ₂ CO 4 _{0,4} → 3 _{0,3}	290623.405	20.96	5.9 ± 0.4
H ₂ CO 4 _{2,3} → 3 _{2,2}	291237.767	68.09	2.4 ± 0.3
H ₂ CO 4 _{3,2} → 3 _{3,1}	291380.488	126.95	3.4 ± 0.3
H ₂ CO 4 _{3,1} → 3 _{3,0}	291384.264	126.95	3.7 ± 0.3
H ₂ CO 4 _{1,3} → 3 _{1,2}	300836.635	33.45	9.8 ± 0.5
H ₂ CO 5 _{1,5} → 4 _{1,4}	351768.645	45.57	10.1 ± 0.9
CH ₃ OH- <i>E</i> 6 ₀ → 5 ₀	289939.477	47.93	0.5 ± 0.1
CH ₃ OH- <i>E</i> 6 ₋₁ → 5 ₋₁	290069.824	40.39	0.9 ± 0.2
CH ₃ OH-A 6 ₀ → 5 ₀	290110.666	34.82	1.2 ± 0.2
CH ₃ OH- <i>E</i> 6 ₁ → 5 ₁	290248.762	55.87	0.3 ± 0.2
CH ₃ OH- <i>E</i> 6 ₂ → 5 ₂	290307.376	60.72	0.3 ^a ± 0.2
CH ₃ OH- <i>E</i> 6 ₋₂ → 5 ₋₂	290307.643	57.07	
CH ₃ OH-A 1 ₁ → 1 ₀	303366.890	2.32	0.6 ± 0.2
CH ₃ OH-A 2 ₁ → 2 ₀	304208.350	6.96	0.8 ± 0.2
CH ₃ OH-A 3 ₁ → 3 ₀	305473.520	13.93	0.9 ± 0.2
CH ₃ OH-A 4 ₁ → 4 ₀	307165.940	23.21	1.4 ± 0.3
C ³⁴ S 6 → 5	289209.068	34.70	4.6 ± 0.7
C ³³ S 6 → 5	291485.935	34.98	1.6 ± 0.3
CS 6 → 5	293912.087	35.27	40.7 ± 2.2
CS 7 → 6	342882.850	49.37	33.1 ± 1.9
HCS ⁺ 7 → 6	298690.453	43.01	0.3 ± 0.1
HCS ⁺ 8 → 7	341350.229	57.34	0.6 ± 0.1
SO 7 ₆ → 6 ₅	296550.064	50.66	3.7 ± 0.5
SO 7 ₇ → 6 ₆	301286.124	56.50	4.0 ± 0.6
SO 7 ₈ → 6 ₇	304077.844	47.55	7.2 ± 1.3
SO 8 ₇ → 7 ₆	340714.155	64.89	3.6 ± 0.7
SO 8 ₈ → 7 ₇	344310.612	70.96	3.0 ± 1.1
SO 8 ₉ → 7 ₈	346528.481	62.14	7.2 ± 1.2
SO ⁺ 2Π _{1/2} 13/2 → 11/2 <i>e</i>	301361.501	38.92	0.6 ± 0.1
SO ⁺ 2Π _{1/2} 13/2 → 11/2 <i>f</i>	301736.791	39.03	0.4 ± 0.1
H ₂ CS 9 _{1,9} → 8 _{1,8}	304307.645	71.60	0.6 ± 0.1
H ₂ CS 10 _{0,10} → 9 _{0,9}	342946.335	74.13	0.4 ± 0.1
H ₂ CS 10 _{2,9} → 9 _{2,8}	343322.111	126.83	0.4 ± 0.1
H ₂ CS 10 _{1,9} → 9 _{1,8}	348534.225	88.47	0.8 ± 0.2
NS 2Π _{1/2} 13/2 → 11/2 <i>e</i>	299700.097	39.82	0.4 ± 0.1
NS 2Π _{1/2} 13/2 → 11/2 <i>f</i>	300098.057	39.93	0.4 ± 0.1
NS 2Π _{1/2} 15/2 → 13/2 <i>e</i>	345823.532	54.20	0.7 ± 0.2
NS 2Π _{1/2} 15/2 → 13/2 <i>f</i>	346220.774	54.33	0.8 ± 0.2
H ₂ S 3 _{3,0} → 3 _{2,1}	300505.560	154.48	0.5 ± 0.1

^a Blended with the 6₋₂ → 5₋₂ CH₃OH-*E* line.

H₂ column density (9×10^{22} cm⁻²) was derived from our observations of the C¹⁷O(3–2) line, assuming ¹⁷O/¹⁶O ~ 1790 (Wilson & Rood 1994) and $T_{\text{rot}} = 70$ K. The methanol column density then corresponds to a fractional abundance averaged over the beam of 10⁻⁹, typical of dark clouds. The 3σ confidence level for T_{kin} ranges between 50 and 75 K, and $(5-20) \times 10^6$ cm⁻³ for n_{H_2} .

The parameters derived from H₂CO differ significantly from the ones of CH₃OH, as high temperature ($T \sim 150$ K) and moderate density ($n \sim 5 \times 10^5$ cm⁻³) are needed to reproduce the data. Since all H₂CO lines are optically thin, we derived a beam-averaged column density of 9×10^{13} cm⁻² for ortho-H₂CO and 3×10^{13} cm⁻² for para-H₂CO, corresponding to a fractional abundance of 10⁻⁹. Fuente et al. (1996) found the same value, although at another position in the Bar. Our analysis infers a T_{kin} in agreement with the results of Batrla & Wilson (2003) from NH₃, but higher than the one derived by Hogerheijde et al. (1995), 85 ± 30 K, also with formaldehyde, but toward a different position. The density derived is substantially higher than that proposed by Hogerheijde et al. (1995) for the interclump

² <http://www.cdms.de>

³ <http://spec.jpl.nasa.gov>

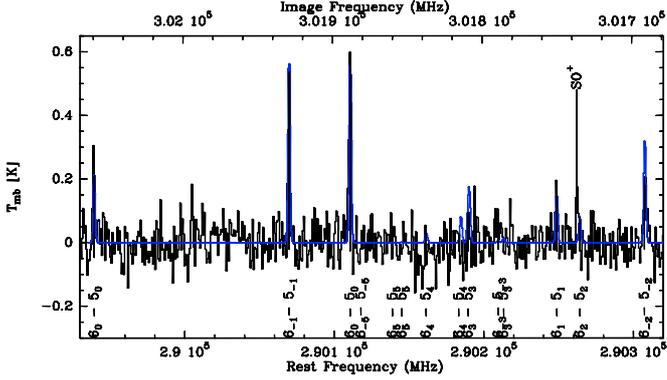


Fig. 1. Spectrum of the $6_k \rightarrow 5_k$ CH_3OH band. Overlaid on the data, in blue, is the synthetic spectrum corresponding to our best fit.

medium ($3 \times 10^4 \text{ cm}^{-3}$), but it agrees with the value inferred by Simon et al. (1997) based on CN. While the n_{H_2} is well-constrained ($1\text{--}5 \times 10^5 \text{ cm}^{-3}$), the temperature we derived is a lower limit.

The infrared pumping through an internal radiation field would probably affect our results for both molecular species. However, since no model of the IR field is available for the Bar, no internal radiation field was taken into account.

Our data suggest that CH_3OH is found mainly in the clumpy medium, while H_2CO traces the interclump material. Both methanol and formaldehyde can form on grain surfaces (Watanabe & Kouchi 2002). It therefore seems remarkable that they trace different environments. We speculate that in the interclump gas, methanol, once released from grain surfaces, is photodissociated to form formaldehyde (one of the possible photodissociation products, Le Teuff et al. 2000). Alternatively, our observations may indicate a different formation mechanism for the two species, surface chemistry for CH_3OH (as no gas-reaction can efficiently form it, Luca et al. 2002), and gas phase chemistry for H_2CO (via the neutral–neutral reaction $\text{CH}_3 + \text{O}$, e.g. Le Teuff et al. 2000). More detailed modeling, including release of molecules from grain surfaces in the PDR, is clearly needed.

3.2. Sulfur-bearing species

The excitation analysis of the S-bearing molecules listed in Table 1 was performed by fitting all the observed transitions with a synthetic spectrum computed under the LTE assumption, in the manner described by Comito et al. (2005). Since the kinetic temperature, the source size, and the column density are degenerate parameters in the optically thin limit, we fixed T_{kin} to the value derived from H_2CO and fit the beam-averaged column density of each species. For SO_2 , OCS, and HCS, upper limits to the column densities were derived, based on the non-detections of lines in our survey range. Results are given in Table 2. The uncertainties on the column densities are derived with a χ^2 analysis and correspond to the 3σ confidence level. CS and its isotopologs are not included in our analysis, because of the anomalous $^{33}\text{S}/^{34}\text{S}$ ratio of ~ 3 we observed (for comparison, see Chin et al. 1996). Observations of other rotational transitions are needed to constrain this ratio reliably.

Sternberg & Dalgarno (1995, hereafter SD95) studied the chemistry of PDRs produced in a dense molecular cloud exposed to intense far-UV radiation fields. The density they employ ($n_{\text{H}} = 10^6 \text{ cm}^{-3}$) is higher than the one relevant here for the

Table 2. Column densities of S-bearing molecules.

Species	$N [\text{cm}^{-2}]$	Species	$N [\text{cm}^{-2}]$
NS	$(1.3 \pm 0.3) \times 10^{13}$	SO_2	$< 3 \times 10^{13}$
SO	$(1.0 \pm 0.2) \times 10^{14}$	HCS	$< 4 \times 10^{14}$
SO^+	$(2.0 \pm 0.5) \times 10^{13}$	OCS	$< 4 \times 10^{13}$
H_2S	$(2.5 \pm 1.0) \times 10^{13}$		
H_2CS	$(2.5 \pm 0.4) \times 10^{13}$		
HCS^+	$(2.5 \pm 0.7) \times 10^{12}$		

interclump gas; discrepancies in the results may be due to this. Their chemical network does not include NS and H_2CS , but all the other S-bearing molecules observed here are found there.

According to SD95, the SO^+/SO ratio (observed 0.2 from Table 2) can be used as diagnostic of the different chemical zones in the cloud and changes between 0.23 at $A_{\text{V}} = 1.5$ and 0.0028 at $A_{\text{V}} = 3$. Assuming that the Bar is edge-on, based on the distance between the ionization front and our position ($\sim 30''$), and assuming an H_2 density of 10^5 cm^{-3} , we derived a visual extinction of ≥ 10 mag (Frerking et al. 1982). Given the complex geometry of the Orion Bar, which has edge-on (but oblique) and face-on parts (Hogerheijde et al. 1995), it is likely that together with the molecular material at this extinction, a hotter, more diffuse layer of gas, corresponding to a face-on surface layer, contributes to the emission we observe. SD95 predict SO_2 to be more abundant than SO and SO^+ at high A_{V} , while our upper limit indicates that SO_2 is less abundant. Similar to the result found for SO^+/SO , our estimates of SO_2/SO and SO_2/SO^+ suggest that these species have a significant contribution from a hotter layer of gas with lower extinction than the molecular layer.

Fuente et al. (2003) report different column densities for SO^+ and SO_2 ; however, their SO^+/SO_2 ratio ranges between 1.2 at the IR front and 0.4 at ($20''$, $-20''$) from it, which is similar to our value of 0.3. Jansen et al. (1995) found values of $N(\text{H}_2\text{S})$ between $8.9 \times 10^{12} \text{ cm}^{-2}$ and $3.9 \times 10^{14} \text{ cm}^{-2}$ across the Bar, with our value of $1.3 \times 10^{13} \text{ cm}^{-2}$ lying in between. Discrepancies in the absolute values of the column densities may depend on the source sizes, on temperatures, and on the total H_2 column density at the observed position.

4. Detection of a deuterated molecule: DCN

Molecular deuteration studies have been extremely powerful in probing the physical history of sources. The deuterium isotopic ratio in molecules can indeed be enhanced in low-temperature environments compared to the cosmic D/H ratio ($\sim 10^{-5}$, Linsky 2003). This enhancement proceeds initially from the transfer of deuterium from the main reservoir HD to two reactive ions (H_2D^+ and CH_2D^+) through exothermic reactions with H_3^+ and CH_3^+ (Roberts & Millar 2000, and references therein). The endothermicity of the backward reactions (respectively by 220 K and 370 K) ensures that the $\text{H}_2\text{D}^+/\text{H}_3^+$ and $\text{CH}_2\text{D}^+/\text{CH}_3^+$ ratios are significantly enhanced at low temperatures. Deuterium is then channeled to other molecules by gas-phase reactions with these two ions and by grain surface reactions. Highly-deuterated molecules can also be found in hot gas, where they are then out-of-equilibrium fossils of an earlier cold phase (e.g. in hot corinos around low-mass protostars, where they are believed to be remnants of the prestellar cold and depleted chemistry; Parise et al. 2004, 2006).

The DCN(4-3) transition at 289.645 GHz ($E_{\text{up}} = 34.56 \text{ K}$) was detected in our survey, and found to be very bright (Fig. 2).

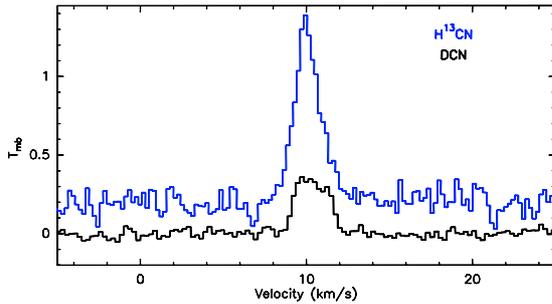


Fig. 2. $\text{H}^{13}\text{CN}(4-3)$ and $\text{DCN}(4-3)$ lines. For clarity, the H^{13}CN line has been translated by 0.2 along the y -axis.

This detection, together with the detection of DCO^+ in the Horsehead nebula (Pety et al. 2006, in preparation), is opening up the study of fractionation processes in PDRs. The $\text{H}^{13}\text{CN}(4-3)$ transition ($E_{\text{up}} = 41.76$ K) was also detected at 345.340 GHz. The H^{13}CN emission was shown from interferometric observations (LS03) to be mostly associated with clumps in the PDR. The DCN emission is thus very likely to also be associated with clumps, as DCN would not survive the high temperatures of the interclump gas. The linewidths of $\text{DCN}(4-3)$ and $\text{H}^{13}\text{CN}(4-3)$ are comparable and non-Gaussian profiles are detected in both cases. We thus assume that H^{13}CN and DCN trace the same region. As the energies of the two lines are very similar, the derived abundance ratio is nearly independent of the temperature. Assuming LTE and a $^{12}\text{C}/^{13}\text{C}$ ratio of 70, we derived beam-averaged DCN and H^{13}CN column densities between 8.4×10^{11} and $1.7 \times 10^{12} \text{ cm}^{-2}$, and between 1.7×10^{12} and $2.8 \times 10^{12} \text{ cm}^{-2}$, respectively, in the range of temperatures 30–150 K. This corresponds to a DCN/HCN ratio of 0.7–0.9%, which is an intermediate value between that observed in warm gas (Orion Hot Core, 0.1%, Schilke et al. 1992), on the one hand, and in dark clouds (L134, 5%, Turner 2001) or cold (30–50 K) gas of the OMC1 ridge region (1–6% Schilke et al. 1992), on the other.

In hot cores/corinos, where the gas has not yet had time to return to equilibrium (which would require timescales of $\sim 10^4$ yr), abundant deuterated molecules can be considered as remnants of earlier cold prestellar chemistry, most probably stored in the ices on grain surfaces and then released when the protostar heats its surroundings. In contrast, the gas in the shielded clumps of the bar might be in steady state. In this case, the observed deuterated molecules have to be present-day product molecules. The detection of DCN in this relatively warm environment, where H_2D^+ would not survive, is in this sense quite striking and is a strong hint of the efficiency of the fractionation

processes occurring via the CH_2D^+ channel, which survives higher temperatures than H_2D^+ , due to the higher exothermicity of the CH_3^+ reaction with HD (Turner 2001). The Orion Bar thus appears a unique reference to test the fractionation reactions involving CH_2D^+ . Knowing the spatial distribution of the DCN/HCN isotopic ratio would then be an interesting tool for understanding the effect of extinction on the fractionation processes.

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